A New α-Haloglycine Template for the Asymmetric Synthesis of Amino Acid Derivatives

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Abstract: Reaction of (S)-N,N-diacylvalylglycine anhydride with N-bromosuccinimide afforded the corresponding α -bromoglycine derivative, which reacted diastereoselectively with allyltributyltin and deuterium over palladium chloride to give the corresponding α -allyl- and α -deuterio-glycine derivatives, respectively.

There have been many reports of the use of α-hąloglycine derivatives in the asymmetric synthesis of amino acids. ¹⁻³ Our recent observation of the selective halogenation of glycine residues in amino acid derivatives resulted in a complementary, though only modestly diastereoselective, method for the elaboration of glycine residues in dipeptides. Thus, for example, reactions of the valyl-α-bromoglycine derivative (1a) with allyltributyltin and diethylmalonate anion gave the dipeptide derivatives (1b) and (1c), respectively, each as a 3:1 mixture of diastereomers. Anticipating that the geometrical constraints imposed by a cyclic system would lead to a greater degree of asymmetric induction, we have now studied analogous reactions of glycine residues in cyclic dipeptides. Prior to this study it had been shown that bis-lactim ethers, such as (2), derived from cyclic dipeptides, react with a high degree of asymmetric induction, although they suffer the disadvantage that they are unstable, decomposing by hydrogen chloride elimination with aromatization to give the corresponding pyrazine derivatives. ⁶

The diketopiperazine (3b), 7,8 chosen for this investigation, was obtained by treatment of valylglycine anhydride (3a) with excess acetic anhydride. Subsequent treatment with N-bromosuccinimide in CCl_4/CH_2Cl_2 (1:1) at reflux under nitrogen, with azobisisobutyronitrile to initiate the reaction, gave the α -bromoglycine derivative (4) [1 H n.m.r. (CDCl₃) δ 0.99 (d, J 7 Hz, 3H), 1.19 (d, J 7 Hz, 3H), 1.85 (m, 1H), 2.61 (s, 3H), 2.63 (s, 3H), 5.08 (d, J 10.5 Hz, 1H) and 6.92 (s, 1H)] as a single diastereomer, in 86% yield. The regioselectivity of this reaction is consistent with the selectivity for reaction of glycine residues observed with acyclic peptides.

When the bromide (4) was treated with allyltributyltin in benzene at reflux for 6 h, with azobisisobutyronitrile to initiate the reaction, a single diastereomer of the corresponding allylglycine derivative (5a) [¹H n.m.r. (CDCl₃) 8 0.82 (d, J 7 Hz, 3H), 1.00 (d, J 7 Hz, 3H), 2.08 (m, 1H), 2.42 (s, 3H), 2.47 (s, 3H), 2.75 (m, 2H), 4.76 (d, J 4.5 Hz, 1H), 4.90 (dd, J 3.5 and 5.5 Hz, 1H), 5.15 (m, 2H) and 5.60 (m, 1H)] was obtained in 60% yield. Although this material proved to be unsuitable for

PhCONH—CH—CONH—CH—CO₂Me MeO

(1)

(1)

(1)

(2)

(3)

(3)

$$R = R$$

(4)

 $R = R$

(5)

 $R = CH_2CH = CH_2$

(6)

 $R = CH(CO_2Et)_2$

(7)

(8)

 $R = R$

(9)

 $R = R$

(1)

 $R = R$

(1)

 $R = R$

(2)

 $R = R$

(3)

 $R = R$

(4)

 $R = R$

(5)

 $R = CH_2CH = CH_2$

(6)

 $R = CH_2CH = CH_2$

(7)

(8)

 $R = R$

(9)

 $R = R$

(1)

 $R = R$

(1)

 $R = R$

(2)

(2)

(3)

 $R = R$

(4)

 $R = R$

(5)

 $R = CH_2CH = CH_2$

(6)

(7)

 $R = CH_2CH = CH_2$

(8)

(9)

 $R = CH_2CH = CH_2$

(1)

(1)

 $R = R$

(2)

(2)

(3)

 $R = R$

(4)

 $R = R$

(5)

 $R = CH_2CH = CH_2CH$

(6)

(7)

(8)

 $R = R$

(9)

 $R = R$

(1)

(1)

X-ray crystallographic analysis, on standing in moist ethyl acetate/light petroleum it underwent hydrolysis to give the monoacyldiketopiperazine derivative (5b) [1 H n.m.r. (CDCl₃) δ 1.00 (d, J 7 Hz, 3H), 1.08 (d, J 7 Hz, 3H), 2.10 (m, 1H), 2.54 (s, 3H), 2.80 (m, 2H), 4.18 (dd, J 3.5 and 9 Hz, 1H), 4.89 (dd, J 1.5

and 8.5 Hz, 1H), 5.30 (m, 2H) and 5.70 (m, 1H)], which was shown to be the *trans*-isomer by crystallographic analysis (Figure 1).9 It is logical to assume that the stereochemistry of this material is the same as that of the diacyldiketopiperazine derivative (5a).

In the presence of Eu(hfc)₃, ¹⁰ the ¹H n.m.r. spectrum of the diketopiperazine (5a) displayed distinct resonances for each enantiomer. In particular, duplicate signals were observed for one of the methyl groups of the isopropyl substituent and for each N-acetyl group. When the above reactions were repeated with the enantiomer of the

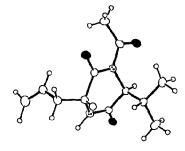


Figure 1. Molecular structure of the monoacyldiketopiperazine derivative (5b)

diketopiperazine (3b) derived from (S)-valine, ¹¹ only one enantiomer of the allylglycine derivative (5a) could be detected using this procedure. ¹² Thus the elaboration of the glycine derivative (3b) occurs without racemization of the valine residue, and the homochiral (3S,6R)-diastereomer of the allylglycine derivative (5a) is produced under these circumstances.

The bromide (4) in THF/D₂O (4:1) was stirred over palladium chloride under an atmosphere of deuterium, at room temperature for 14 h, to give the deuteride (6a), with 92% deuterium incorporation, as a 20:1 mixture of diastereomers. That mixture was treated with hydrazine hydrate in DMF¹³ to give the deuteriated valylglycine anhydride (6b), also as a 20:1 mixture of diastereomers. In the ¹H n.m.r. spectrum (D₂O) of the latter compound, the relative intensity of the resonances at δ 4.20 and 4.37, for the α -hydrogen of the glycine residue in the *cis*- and *trans*-isomers, respectively, ¹⁴ showed the *cis*-diastereomer to be predominant. Through correlation, the major diastereomer of the diacyldiketopiperazine derivative (6a) can also be assigned the *cis*-stereochemistry.

Although the relative stereochemistry of the bromide (4) was not separately determined, it is likely that the bromine is incorporated *trans* to the isopropyl substituent. The conversion of the bromide (4) to the deuteride (6a) would then involve an inversion of configuration.

Treatment of the bromide (4) with tributyltin deuteride⁵ gave the deuteriated diketopiperazine (6a), with 85% deuterium incorporation, as a 2:1 mixture of the *trans*- and *cis*-diastereomers. The predominance of the *trans*-isomers of the diketopiperazines (5a) and (6a), in the reactions of the bromide (4) with allyltributyltin and tributyltin deuteride, respectively, can be attributed to a preference for delivery of deuterium and the allyl group *trans* to the isopropyl group in the radical derived by bromine atom transfer from the halide (4).

It is noteworthy that, by analogy with the present work, the palladium-catalysed reaction of the bromide (7) with deuterium has been found to be highly stereoselective, affording mainly the deuteride (8) through inversion of configuration.² The reaction of the bromide (7) with tributyltin deuteride was much less stereoselective and gave mainly the deuteride (8) resulting from retention of configuration.²

The synthesis of the bromide (4) and its subsequent reactions to give the allylglycine derivative (5a) and the deuteriated diketopiperazine (6a) illustrate the high degree of diastereoselectivity that can be expected in the elaboration of α -haloglycine derivatives in cyclic dipeptides. On this basis, the bromide (4) is likely to have considerable utility as a template for the asymmetric synthesis of amino acid derivatives. In this regard it should be noted that both (R)- and (S)-valine are inexpensive and readily available, therefore the approach delineated above should prove suitable for the α -substitution of glycine residues to give whichever enantiomer of the product is desired.

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References and Notes

- For examples see: Kober, R.; Papadopoulos, K.; Miltz, W.; Enders, D.; Steglich, W.; Reuter, H.; Puff, H. Tetrahedron 1985, 41, 1693; Emert, P.; Meyer, J.; Stucki, C.; Schneebeli, J.; Obrecht, J.-P. Tetrahedron Lett. 1988, 29, 1265; Belokon, Y. N.; Popkov, A. N.; Chernoglazova, N. I.; Saporovskaya, M. B.; Bakhmutov, V. I.; Belikov, V. M. J. Chem. Soc., Chem. Commun. 1988, 1336; Hamon, D. P. G.; Razzino, P.; Massy-Westropp, R. A. J. Chem. Soc., Chem. Commun. 1991, 332; Hamon, D. P. G.; Razzino, P.; Massy-Westropp, R. A. J. Chem. Soc., Chem. Commun. 1991, 722; Sinclair, P. J.; Zhai, D.; Reibenspies, J.; Williams, R. M. J. Am. Chem. Soc. 1986, 108, 1103; Williams, R. M.; Sinclair, P. J.; Zhai, D.; Chen, D. J. Am. Chem. Soc. 1988, 110, 1547; Williams, R. M.; Sinclair, P. J.; Zhai, W. J. Am. Chem. Soc. 1988, 110, 482; Zhai, D.; Zhai, W.; Williams, R. M. J. Am. Chem. Soc. 1988, 110, 2501.
- 2. Williams, R. M.; Zhai, D.; Sinclair, P. J. J. Org. Chem. 1986, 51, 5021.
- 3. For a review see: Williams, R. M. Synthesis of Optically Active α-Amino Acids, Organic Chemistry Series, Vol. 7, Baldwin, J. E.; Magnus, P. D. Eds., Pergamon Press, 1989.
- Easton, C. J.; Hay, M. P. J. Chem. Soc., Chem. Commun. 1986, 55; Burgess, V. A.;
 Easton, C. J.; Hay, M. P. J. Am. Chem. Soc. 1989, 111, 1047.
- 5. Easton, C. J.; Scharfbillig, I. M.; Tan, E. W. Tetrahedron Lett. 1988, 29, 1565.
- Schollkopf, U.; Neubauer, H.-J.; Hauptreif, M. Angew. Chem., Int. Ed. Engl. 1985, 24, 1066;
 Schollkopf, U.; Gruttner, S.; Anderskewitz, R.; Egert, E.; Dyrbusch, M. Angew. Chem., Int. Ed. Engl. 1987, 26, 683.
- 7. All new compounds were fully characterized.
- 8. Initial reactions and the X-ray crystallographic structure determination of the diketopiperazine (5b) were performed using racemic materials.
- 9. Molecular structure of the diketopiperazine (5b): monoclinic space group $P2_1/c$, a = 8.778(2), b = 15.416(1), c = 10.012(3) Å, $\beta = 109.24(1)^0$, R = 0.050 for 1068 reflections.
- Kessler, H.; Molter, M. Angew. Chem., Int. Ed. Engl. 1973, 12, 1011; Kessler, H.; Molter, M. J. Am. Chem. Soc. 1976, 98, 5969.
- 11. Kanmera, T.; Lee, S.; Aoyagi, H.; Izumiya, N. Int. J. Peptide Protein Res. 1980, 16, 280.
- 12. On this basis the e.e. was deemed to be >95%.
- 13. Fujinaga, M.; Matsushima, Y. Bull. Chem. Soc. Jpn. 1966, 39, 185.
- 14. Davies, D. B.; Khaled, M. A. J. Chem. Soc., Perkin Trans. 2 1976, 1238.